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ASPECTS OF ENVIRONMENTAL AND INDUSTRIAL SAFETY OF NON-AQUEOUS PROCESS FLUIDS IN CONSTRUCTION AND COMPLETION OF WELLS

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АСПЕКТЫ ЭКОЛОГИЧЕСКОЙ И ПРОМЫШЛЕННОЙ БЕЗОПАСНОСТИ ПРИМЕНЕНИЯ ТЕХНОЛОГИЧЕСКИХ ЖИДКОСТЕЙ НА НЕВОДНОЙ ОСНОВЕ В ПРОЦЕССАХ СТРОИТЕЛЬСТВА И ОСВОЕНИЯ СКВАЖИН

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Nowdays, reserves complex in terms of geology, pressure and temperature are involved into production. That is accompanied by the increase in use of non-aqueous process fluids, especially invert-emulsion drilling muds. One of the key problems that are an obstacle for mass application of non-aqueous fluids in construction and completion of wells is a high man-made effect on the environment due to their environmental hazard and need to ensure enhanced industrial safety measures when used in field conditions. The classification of dispersion media used in modern non-aqueous process liquid formulations is proposed. The following indicators of "environmental friendliness" and industrial safety of this type of liquids are considered: toxic effects on ecosystems, ability to degrade in the natural environment to safe products under the influence of microorganisms, indicators of industrial sanitation and fire safety. The factors affecting the flash point temperature of non-aqueous process fluids were analyzed. Stricter requirements for environmental readings of the dispersion medium of process fluids on a non-aqueous basis will improve the environmental situation in areas of drilling operations. Knowing the basics of management of fire-hazardous properties of this type of liquids will increase the industrial safety of their use in field conditions. The aspects of environmental and industrial safety of process fluids on a non-water basis, presented in the paper, will be useful for specialists in the field of drilling and production who are engaged in design and development of process fluids, including shale oil and gas deposits.

Ключевые слова:

технологические жидкости на
неводной основе, инвертно-
эмульсионные буровые
растворы, олефины, парафины,
степень биодegradации, класс
опасности, токсическое
воздействие на экосистемы,
температура вспышки.

В настоящее время в процессах строительства и освоения нефтегазовых скважин в связи с вовлечением в разработку месторождений со сложными геологическими и термобарическими условиями наметилась отчетливая тенденция роста объемов использования технологических жидкостей на неводной основе, в особенности инвертно-эмульсионных буровых растворов. Одной из ключевых проблем, являющихся препятствием для массового внедрения жидкостей на неводной основе в процессы строительства и освоения скважин, является высокая техногенная нагрузка на окружающую природную среду вследствие их экологической опасности, а также необходимость обеспечения повышенных мер промышленной безопасности при их применении в промысловых условиях. Предложена классификация дисперсионных сред, используемых в современных рецептурах технологических жидкостей на неводной основе. Рассмотрены следующие показатели экологичности и промышленной безопасности данного типа жидкостей: токсическое воздействие на экосистемы, способность к деградации в природной среде до безопасных продуктов под воздействием деятельности микроорганизмов, показатели промышленной санитарии и пожаробезопасности. Исследованы факторы, влияющие на значение температуры вспышки технологических жидкостей на неводной основе. Ужесточение требований в отношении экологических показателей дисперсионной среды технологических жидкостей на неводной основе позволит улучшить экологическую ситуацию в районах ведения буровых работ. Знание основ управления пожароопасными свойствами данного типа жидкостей позволит повысить промышленную безопасность их применения в промысловых условиях. Представленные в статье аспекты экологической и промышленной безопасности технологических жидкостей на неводной основе будут полезны специалистам в области бурения и добычи, занимающимся проектированием и разработкой технологических жидкостей, в том числе для сланцевых месторождений нефти и газа.

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Introduction

Nowdays, reserves complex in terms of geology, pressure and temperature are involved into production. That is accompanied by the increase in use of non-aqueous process fluids, especially invert-emulsion drilling muds. Currently, all leading foreign and domestic service companies have this type of drilling mud [1]. Due to the ability to preserve natural reservoir properties non-aqueous process fluids are also widely used as killing, completion and perforation fluids [2, 3]. The main advantage of using this type of killing fluids and perforation is to prevent swelling and dispersion of formation rocks composed of clay minerals. [4].

One of the key problems that are an obstacle for mass application of non-aqueous fluids in construction and completion of wells is a high man-made effect on the environment due to their environmental hazard and need to ensure enhanced industrial safety measures when used in field conditions.

Unit recently natural hydrocarbon liquids such as oil and gas condensates and oil refining products (kerosene, gas oil, and diesel fuel) have been used as the most common dispersion medium for non-aqueous emulsion drilling fluid (NAEDF) tested in practice [5-7]. Availability and relatively low price were the main features for them to be used as a dispersion medium of non-aqueous process fluids. Oil and its products contain compounds that are additional natural stabilizers and emulsifiers for inverse emulsions in terms of their properties. Therefore, obtaining highly stable structured solutions on their basis is a fairly simple technological task. These compounds mainly belong to the group of aromatic, oxygen-containing and sulfur compounds (resins, asphaltenes). At the same time, it is aromatic compounds that determine the high toxicity of hydrocarbon-based solutions [8]. According to

the degree of harmful effects on ecosystems, oil and oil products rank second after radioactive contamination [9].

Classification of dispersion media of process liquids

In recent years, due to the fact that environmental requirements have become tightened, it is necessary to replace traditional hydrocarbon liquids (oil, diesel fuel) with more environmentally friendly fireproof biodegradable hydrophobic fluids while retaining all the advantages of emulsion drilling fluids. Currently, the development of fire and environmentally hazardous NAEDF modifications goes in several directions. The following products are used as the basis of invert emulsions [10-15]:

- refined and less toxic mineral oils that do not contain high concentrations of mono- and polyaromatic hydrocarbons that are toxic to biota;
- isolated from the mixture of hydrocarbons separate fractions (mainly paraffin-naphthenic), subjected to biodegradation;
- products based on vegetable raw materials.

The classification of dispersion media used in modern NAEDF formulations [16] is shown in Fig. 1

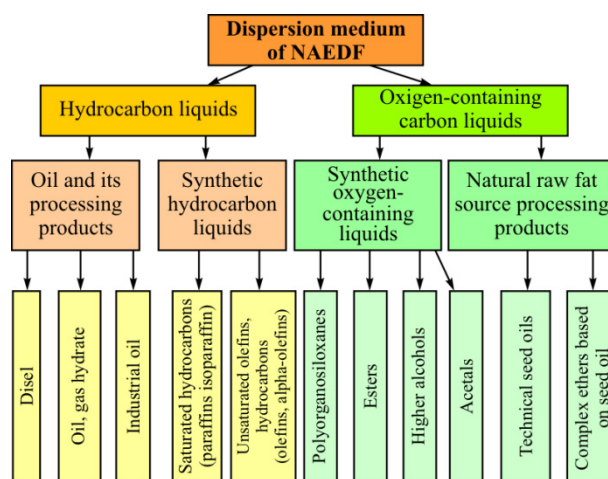


Fig. 1. The classification of dispersion media NAEDM

Table 1

Synthetic oils used to prepare non-aqueous process fluids [22-24]

Tradename	Synthetic base solution	Manufacturer
ISO-TEQ	Olefins (mixture C ₁₄ -C ₂₀)	Baker Hughes INTEQ
NX-3500	Ester	Baker Hughes INTEQ
BG-550	Ester	Baker Hughes INTEQ
PT-3500	Isoparaffin	Baker Hughes INTEQ
ALPHA-TEQ	Linear alpha olefins (mixture C ₁₄ -C ₁₆)	Baker Hughes INTEQ
NOVAPLUS	Olefins (mixture C ₁₆ -C ₁₈)	Schlumberger
ECOGREEN	Ester	Schlumberger
NOVATEC	Linear alpha olefins (mixture C ₁₄ -C ₁₈)	Schlumberger
PARADRIL	Paraffin	Schlumberger
PETROFREE SF	Olefins (mixture C ₁₆ -C ₁₈)	Baroid
PETROFREE	Ester	Baroid
PETROFREE LE	Linear alpha olefins (mixture C ₁₄ -C ₁₈)	Baroid
XP-07	Linear Paraffin	Baroid

Application of ethers, polyorganosiloxanes, polyalkylene glycols, polyalphaolefins is considered in many solutions of the modern process fluids [17-21]. Operators engaged in offshore drilling back in the 1980s last century began research to create synthetic-based drilling muds that minimize damage to the marine environment. In 1993, a well was drilled in the North Sea using M-I Drilling Fluids based on polyalphaolefins, hydrocarbons with a chain length of about 10-12 atoms [18]. Types of synthetic fluids used in the formulations of leading foreign companies are given in Table 1. Application of synthetic oils reduces the load on the environment, expands the possibilities of using oils, including as the basis of emulsion drilling fluids, in conditions of both negative and high temperatures. However, the high cost and narrowness of the domestic range of synthetic oils limit their use in drilling technologies.

Vegetable oils (peanut, soybean, linseed, corn, castor etc.) and their products are ones of the most ecologically safe organic substances [25-29]. The basis of vegetable oil is esters (triglycerides) of high-molecular saturated and unsaturated carboxylic acids. The advantage of vegetable oils is their almost complete biodegradation under aerobic and anaerobic conditions. The lack of invert solutions based on organic liquids of plant origin is the complexity of their preparation and regulation of technological parameters, due to the variability of the composition and properties of

plant materials for the production of oils, high viscosity and pour point of the original oils.

Indicators of environmental safety of non-aqueous process fluids

Environmental friendliness and industrial safety of non-aqueous process fluids can be assessed by the following indicators:

- toxic effects on ecosystems,
- the ability to degrade in the natural environment to safe products under the influence of the activity of microorganisms,
- indicators of industrial sanitation and fire safety.

According to ASTM, toxicity of organic liquids is estimated by the toxicity of their aqueous extracts using various biotesting methods, one of which is 25 for diesel fuel, 11.5 for acetal, 4 for alpha-olefin, 1.6 for coriander oil, and polyalphaolefins for zero [30]. Results of a large number of experiments conducted by the European Center for Ecotoxicology and Toxicology of Chemicals showed that the action of anaerobic bacteria in different liquids is not the same. Biodegradation was found to be very high in fatty acids and essential oils. At the same time, in mineral oils, diethyl ether, polyalphaolefins and a number of other compounds, biodegradation was absent or only partially occurred [31]. In our country, in accordance with the “Criteria for classifying waste as hazardous for the environment” [32] for liquid waste, including

NAEDM, no water extracts are made, and the viability of hydrobionts in the environment of the waste itself is checked (i.e. oils). The calculated method of determining the hazard class for the environment is applicable only to waste of I-IV classes, since it is obligatory to conduct biotesting of the studied waste for class V. Due to the fact that hydrobionts cannot develop normally in an oily environment without access of dissolved oxygen, the assignment of any organic liquid to the V class of hazard by the method of biotesting is impossible.

Many developers of non-aqueous process fluids estimate the environmental safety of oils by the degree of their biodegradation. In special reviews [33-36], it is reported that almost all hydrocarbons to some extent undergo microbiological oxidation. For example, light petroleum products such as diesel fuel at an initial concentration in the soil of 0.5 % in 1.5 months degrade by 10-80 % of the initial amount depending on the content of volatile hydrocarbons. The degree of biodegradation of unsaturated aliphatic hydrocarbons (olefins and alpha-olefins) in aerobic conditions exceeds 70 %, in anaerobic conditions – 53 % [37]. Aromatic hydrocarbons have the highest resistance to microbiological oxidation. Thus, application of paraffins, which are most susceptible to microbial oxidation, and also paraffin-naphthenic fractions of oil and oils as basics of process fluids to a greater extent meet the requirements of environmental safety than the use of oil products with a high content of aromatics [10].

When assessing the ability of a liquid for biodegradation, it is necessary to take into account that microbiological oxidation occurs in the range of 18-40 °C at the optimum of 28-32 °C, which eliminates the process of degradation of any hydrocarbon fluid when it enters the environment (as part of drill cuttings, leaks solution) in the winter and in areas with cold climates [38]. Due to the fact that there is almost no biodegradation in cold climate conditions, currently the legislative bodies of most developed countries, particularly in the UK and Canada during offshore drilling prohibit discharge of spent drilling muds on both

hydrocarbon and synthetic basises. There is Environmental Protection Agency (EPA) in the USA that prohibits the discharge of drilling mud and sludge at a distance of three miles from the coast [39]. Synthetic-based solutions in Norway are being tested for bioaccumulation and biodegradation. Permission to dump waste into the sea is given in the case of favorable results of these tests only. While dumping synthetic-based muds north of the 62nd parallel is prohibited at all [40].

Nowdays, the bodies of Rosprirodnadzor classify the waste that contain 15 % or more of petroleum products as hazard class III substances (moderately hazardous substances). The PND F 16.1:2:2.2:2.3:3.64-10 is the main method for the determination of petroleum products in the calculation of the hazard class of waste 1 [41]. According to this method, the indicator “oil products” is taken as a mixture of non-polar organic substances contained in the waste, soluble in hexane. In addition to traditional petroleum products, this indicator includes synthetic liquids soluble in hexane, in particular polyalphaolefins and even vegetable oils. Thus, on the basis of the waste ranking adopted in our country, all non-aqueous process fluids will be classified as hazard class III. It comes that there is no reason to replace traditional hydrocarbon liquids with their environmentally safe counterparts and indicates the need to revise the methods used to determine oil products to eliminate environmentally friendly organic liquids from this indicator.

Another significant factor limiting the use of non-aqueous process fluids, especially oil, in field conditions, is their high fire safety, evaporation and toxic effects on the human body when it enters through the air passage and skin. The risk to life and health increases significantly with presence of hydrogen sulfide in oil and oil products. The negative impact of mineral oils on the human body occurs when exposed to open skin or when working in clothes soaked with them, as well as inhalation of oily fumes or mist. State-of-the-art synthetic oils on the basis of paraffins are not harmful if swallowed and do not irritate the skin and mucous membranes of a

person. Therefore, they are preferable in terms of meeting industrial safety requirements [42]. Due to predominant use of traditional hydrocarbon products in the construction of wells, there is an urgent need to introduce a wide range of organizational and technical measures to ensure the safety of personnel working directly with this type of fluid.

Investigation of factors affecting the flash point temperature of non-aqueous process fluids

Most organic liquids used as an NAEDF dispersion medium belong to a group of flammable substances that can flash upon contact with an open source of fire. The flammable and explosive properties of organic liquids, which characterize their ability to ignite during heating and subsequent flame spread, determine the flash point. The flashpoint is the lowest temperature at which vapors above the surface of a flammable substance flash upon contact with an open source of fire [43]. There are two methods for determining the flash point of petroleum products in open (GOST 4333-87) and closed (GOST 6356-75) crucibles are standardized. V.N. Glushchenko notes that when controlling the NAEDF flash point, preference should be given to an open crucible. This is due to the simulation of the real conditions of use of drilling fluids in wells [44].

The greatest danger in field conditions are fluids belonging to the category of flammable, i.e. able to flash at temperatures below 61 °C in a closed crucible and below 66 °C in an open one [43]. At the same time, the main danger of using flammable liquids is related to the so-called human factor: non-compliance of safety requirements by the drilling crew when working with such liquids. In particular, when using NAEDF in the fields of the Perm region, a number of cases of flame occurrence above the surface of the hydrocarbon base NAEDF were noted at the stage of solution preparation while simultaneously carrying out welding and smoking personnel.

There are quite a few works on the correlation of the flash point of organic liquids depending on their composition. At the same time, there is almost nothing about the study of factors affecting

the flash point of technological fluids on a non-aqueous basis.

It is obvious that the flash point of anhydrous process liquids corresponds to the flash point of their dispersion medium. According to our assumption, the flash point of emulsion process fluids, including NAEDF, is determined by the flash point of its dispersion medium and water content. In addition, the study the effect on the flash point of solutions of the quantitative content of insoluble solids – weighting agents (calcium carbonate and barite) in them can be of some interest.

For certain classes of hydrocarbons, empirical coefficients a and b are derived. The coefficients connect the flash point with their boiling point according to the following equation [45]:

$$T_{\text{flash}} = a + b \cdot t_{\text{boil}}.$$

Using these coefficients, we obtain the following dependence:

– for saturated hydrocarbons

$$T_{\text{flash}} = 0.693T_{\text{boil}} - 73.22;$$

– for aromatic hydrocarbons

$$T_{\text{flash}} = 0.665T_{\text{boil}} - 67.83.$$

Thus, based on the data presented for the preparation of NAEDF, it is recommended to use diesel fuels, mineral, synthetic and low-viscosity low-caking base oils (the basis of commercial oil without additional additives) containing hydrocarbon fractions with a T_{boil} not lower than 194 °C (according to calculations is more than 61 °C). Data on the maximum solution temperature at the wellhead is required to issue recommendations for high-temperature wells.

Due to the fact that low viscosity hydrocarbon liquids, including diesel (marine) fuel, are the most suitable for the preparation of NAEDF from an economic point of view, it becomes very important to increase their flash point. The results of studies on the possibility of increasing the flash point of flammable by introducing high-boiling hydrocarbon fractions are presented in Table 2.

Data given in Table 2 indicate that the addition of mineral oil at a concentration of 15% can increase the total fuel oil T_{flash} by only 4 °C, while

the kinematic viscosity of the mixture increases by 1.3 times, which leads to an undesirable increase in the rheological properties of NAEDF, especially plastic viscosity. The actual value of the flash point of a mixture of hydrocarbons is always lower than the arithmetic average temperature of the flash points of the components that make up the mixture calculated by the additivity rules. This is due to the fact that the flash point depends mainly on the vapor pressure of the low-boiling component, and the high-boiling component serves only as a heat transmitter.

Table 2

Flash point of hydrocarbon liquids
of different composition

Fluid composition	T_{flash} , °C	Kinematic viscosity at 20 °C
Diesel fuel	53	2.73
Diesel fuel – 95 %, base oil – 5 %	52	2.90
Diesel fuel – 90 %, base oil – 10 %	53	3.21
Diesel fuel – 85 %, base oil – 15 %	57	3.43
Diesel fuel – 95 %, oil I-8 – 5 %	53	3.03
Diesel fuel – 90 %, oil I-8 – 10 %	53	3.49
Diesel fuel – 85 %, oil I-8 – 15 %	55	3.83

Note: T_{flash} of base oil – 163 °C, T_{flash} of oil I-8 – 152 °C, flash point was determined in a closed crucible. The matrix of mineral hydraulic oil without additional additives was used as the base oil.

The second factor determining the flash point of NAEDF is the ratio in the solution of the hydrocarbon and aqueous phases. It is considered a priori that the higher the content in the solution of the aqueous phase, the higher the flash point is. Fig. 2 presents data on the effect of water content on the value of the flash temperature of NAEDF based on diesel fuel.

The most NAEDF formulations of leading foreign companies the water-oil ratio of hydrocarbons is 80:20 and 70:30. Such the insignificant water content allows to increase the flash point of the solution by 15-20 % compared to the flash point of the fuel used for its preparation. This suggests that in order to prepare

NAEDF with low water content it is recommended to use mineral oils having a high flash point.

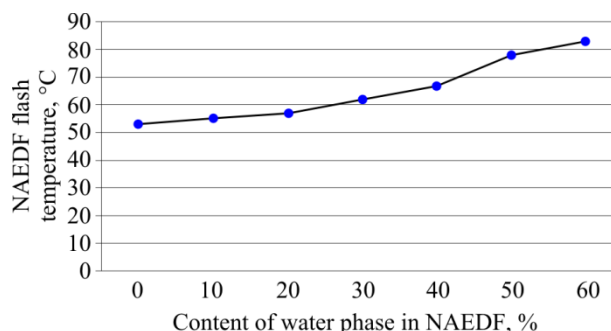


Fig. 2. The effect of the content of the aqueous phase in NAEDF on the flash point of the solution

According to the “Safety Rules in the Oil and Gas Industry” (SROGI), the flash point of a hydrocarbon-based drilling fluid must be 50 °C higher than the maximum temperature at the wellhead [46]. Thus, it becomes very important at the stage of drilling fluid design to predict the final flash temperature of NAEDF considering water content and passport value of the flash point of the oil or fuel used to prepare the solution. In order to identify statistical dependence we conducted laboratory studies to determine the flash temperature of NAEDF using diesel fuel from different producers. It is important to ensure the formulation of the solution has not changed: the content of the aqueous phase (solution of calcium chloride) was 50 vol. %. A total of 30 NAEDF samples were prepared, and as a result of the study and statistical data processing, we obtained the following dependence (Fig. 3). Thus, knowing the passport value of the flash point of the fuel or oil, it is possible to calculate the approximate value of the NAEDF flash point, prepared on the basis of this fluid.

The second goal of the study was to built similar dependencies for NAEDF prepared on the basis of mineral and synthetic oils with a flash point of more than 100 °C. The research results showed that the difference between the flash point of NAEDF and the flash point of the oil is within the measurement error; the presence of the aqueous phase does not affect the flash point of the solutions (Table 3).

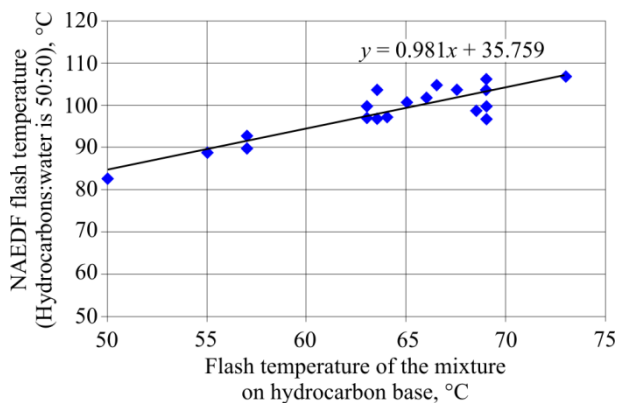


Fig. 3. Dependence of the flash temperature of NAEDF from flash point of diesel fuel

Table 3

Flash point of NAEDF, prepared on the basis of mineral oils with a flash point of more than 100 °C

NAEDF dispersion medium	T_{flash} of dispersion medium, °C	T_{flash} of NAEDF, °C
Industrial oil I-8	152	150
Transformer oil T-1500 U	135	137
Base mineral oil	163	170
Polyalphaolefins PAOM-2	155	157

Note: phase ratio of NAEDF hydrocarbons:water is 50:50.

It is known from theoretical calculations that the flash point of NAEDF depends on the properties of the dispersion medium (oil), but is much higher than the flash point of the latter [47]. Therefore, having obtained such ambiguous data, we carried out repeated experiments, during which similar results were obtained. When the solutions are heated above 120–130 °C the water phase boils away completely corresponding to the flash point of the oil, only the hydrocarbon medium with solid fillers remains in the solution. It became clear only with the analysis of the method for determining the NAEDF flash point associated with heating the solution by electrical heating in an open crucible to temperatures well above the boiling point of water. This conclusion is very important when designing solutions for high-temperature wells, since it allows to predict the flash temperature of NAEDF and choose mineral and synthetic oil as the basis of the solution, the flash point of which corresponds to SROGI.

Usually, NAEDFs are a three-phase system containing weighting agents and builders as a solid phase. The content of the weighting agent in

NAEDF (barite and CaCO_3) can reach a rather high value – up to 50-70 wt. %, due to the need to obtain solutions with the required density. Accordingly, there is a question arised about the effect of the concentration of a solid phase in a solution on its flash point. Our studies have shown that it is not possible to regulate NAEDF flash point by changing the content in the system of the solid phase (Table 4).

Table 4

Influence of barite content on NAEDF flash point

Content of barite in NAEDF, mas. %	Content of aqueous phase in solution, vol. %	T_{flash} of dispersion medium, °C	T_{flash} NAEDF, °C
No	50	63.0	97.0
5	50	63.5	98.0
10	45	67.0	100.5
15	45	69.5	109.0
20	40	73.0	101.0
50	35	72.0	86.0

Weighting agents presented in the solution would not increase the value of its flash point even at high concentrations. As it was mentioned above, this is due to the fact that the flash point of the solution is determined primarily by the flash point of volatile combustible components contained in the hydrocarbon-based solution, the volumetric content of which in the gas-air mixture above the surface of the solution does not depend on solid phase.

Conclusion

There is a trend that shows complication of geological conditions in well drilling. Therefore, the share of non-aqueous process fluids used in drilling will inevitably increase. Knowledge of the basics of managing the fire-hazardous properties of this type of liquids will improve the industrial safety of their use in field conditions. Making requirements stricter in terms of environmental performance of the dispersion medium of non-aqueous process fluids will improve the environmental conditions in drilling areas. The analysis of the environmental performance of various organic hydrophobic liquids indicates only relative safety and low hazard of synthetic and plant products, since the safety of these compounds can be considered and compared only

with respect to oil and toxic products of its processing. These fluids are more fire-resistant, characterized by less evaporation and toxic effects on the human body when entering through the respiratory tract and skin. At the same time, when

working with organic liquids, their discharge into objects of the natural environment is unacceptable. If an organic liquid released on the soil and water bodies, it is necessary to collect and neutralize them by special methods.

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